

# Portable Gas Detection

## Electrochemical Oxygen Sensors and XCell<sup>®</sup> O<sub>2</sub> Sensors



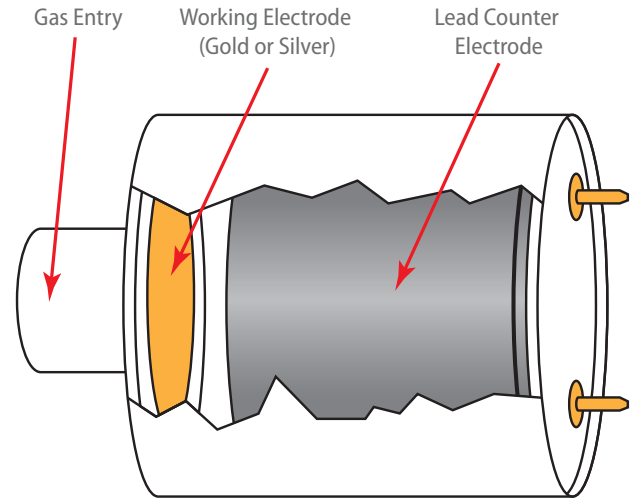
Technical Bulletin

*Electrochemical oxygen sensors detect oxygen deficiency or enrichment in the environment.*

### Standard O<sub>2</sub> Sensor Technology

Oxygen sensors use electrochemical reactions to generate a current proportional to the concentration of oxygen and report the amount of oxygen in percent volume.

Most O<sub>2</sub> sensors on the market use a consumable chemical reaction: oxygen gas reacts with lead to form lead oxide. This chemical reaction is constantly taking place from the time the sensor is manufactured. As lead is converted to lead oxide, the lead is consumed until it is eventually used up. At this point, the sensor's output drops below a point where the sensor will continue to calibrate. When this happens, the sensor reaches end of life. The lifespan of a typical lead based O<sub>2</sub> sensor is about 18-24 months.

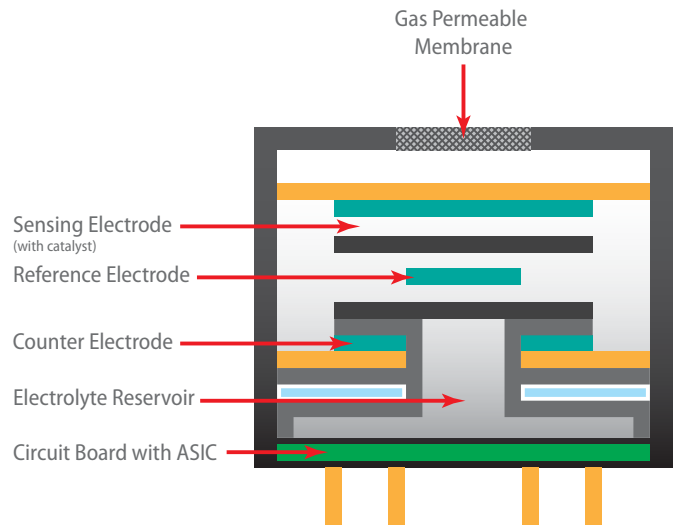


### Why MSA XCell O<sub>2</sub> Sensors?

#### Longer Life!

The electrochemical system in the MSA XCell O<sub>2</sub> Sensor uses a *non-consuming* chemical reaction. O<sub>2</sub> molecules entering the sensor react with the working electrode creating electron flow and water as a byproduct. At the counter electrode, water is converted back into O<sub>2</sub> molecules. The chemical reaction requires a low voltage, which is controlled by the application-specific integrated circuit (ASIC) in the XCell Sensor. Because nothing is consumed or "used up" as the sensor is functioning, the sensor has a longer lifespan. XCell O<sub>2</sub> Sensors have a typical life of more than four years.

This non-consuming chemical reaction also means that the sensor can have a much longer shelf-life. The chemical reaction in a traditional lead-based O<sub>2</sub> sensor starts the second the sensor is manufactured. At this time, the chemical process whereby the lead is converted to lead oxide begins and the finite sensor life begins. The MSA XCell O<sub>2</sub> sensor is designed to have a very long shelf-life. While the sensor sits un-powered, no chemical reactions with O<sub>2</sub> are taking place and no life is being depleted from the sensor.



*Because every life has a purpose...*

### Performance! Response Time—Pressure Sensitivity

The chemical reaction used by the MSA XCell sensor allows the sensor to respond very quickly to changing O<sub>2</sub> levels with a typical t<sub>90</sub> response time of less than 10 seconds.

Most O<sub>2</sub> sensors on the market, including the XCell O<sub>2</sub> sensor, are capillary limited O<sub>2</sub> sensors. This means O<sub>2</sub> is drawn into the sensor by long, thin capillary channel. This capillary keeps the balance between the external ambient pressure and the internal ambient pressure. The ratio of length to diameter of the capillary determines how quickly the sensor equilibrates and how large of a “spike” is caused by quickly changing ambient pressures. The combination of the capillary and the ASIC electronic control gives the XCell O<sub>2</sub> sensor an edge over other sensors by responding to rapid pressure changes. Pressure transients are equalized in seconds and the peak readings from these pressure spikes are minimized to display very small changes for that brief period. This process eliminates unwanted alarms due to pressure changes.

## Things to know about Oxygen Sensors!

- Water vapor in the air displaces other gases in the atmosphere including O<sub>2</sub>. This is why it is more difficult to breathe in very hot/humid conditions. If you suddenly take an instrument from a relatively dry area where you calibrated it into a very humid area, your O<sub>2</sub> sensor could read several tenths of a percent low. This is not the sensor drifting due to humidity; this is an actual reading that the percent O<sub>2</sub> is slightly lower in that area.
- We calibrate our O<sub>2</sub> sensor in a background of Nitrogen, just like the air we breathe. During this calibration, the flux, or rate that O<sub>2</sub> molecules enter the sensor in nitrogen is part of the calibration. If the background gas significantly changes to another inert gas like Helium or Argon, then this rate at which O<sub>2</sub> molecules enter the sensor will change. The sensor will read high if the background gas molecular size is less than nitrogen, such as helium. It will read low if the background gas molecular size is larger than nitrogen, such as argon.
- Lead-free O<sub>2</sub> sensors do not exhibit cross sensitivities to other gases, just O<sub>2</sub>.
- O<sub>2</sub> sensors can be damaged by large amounts of gases that have the ability to damage the platinum working electrode of the sensor. This would typically be percent by volume concentrations of Chlorines or Fluorines (an example would be many refrigerants). If poisoning is suspected, a bump test and/or calibration should be performed.

**Never try to calibrate XCell O<sub>2</sub> sensor immediately after installing a replacement sensor. Install the replacement sensor and wait 20-30 minutes for stabilization before calibrating.**

Note: This Bulletin contains only a general description of the products shown. While uses and performance capabilities are described, under no circumstances shall the products be used by untrained or unqualified individuals and not until the product instructions including any warnings or cautions provided have been thoroughly read and understood. Only they contain the complete and detailed information concerning proper use and care of these products.



**ID 0800-82-MC / October 2013**  
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